



Transition Metals 1: Tungsten, Titanium and Other Elements (The Periodic Table)

Anita Ganeri

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This series uses a common or well-known element to look at the groups of the periodic table and to show the similarities and differences between elements. It uses full-colour illustration of the periodic table and shows the chemical symbol for each element in place, alongside its neighbours. Chemical formulae for common compounds are also shown. Information boxes and tables contain listings of facts and figures. Chemical reactions are interpreted as word equations, and timelines chart the history and discovery of the elements. This title covers the properties of the first section of transition metals in the periodic table (which includes naturally occurring and artificially made elements), the trends that develop across the period and how the slow nature of the chemical reactions compare with other elements. Typical examples are chromium, molybdenum and manganese, which form coloured compounds, make strong alloys with other elements and provide good catalysts.

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